



DNASE I and STRUCTURE CATALYTIC SITE

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ABSTRACT

This Article describes 3Dimention studies carried out of catalytic site over the past year which have advanced of understanding the substrate to recognition and how the catalytic mechanism of the several key DNase I. The hydrolysis reaction that occurs in DNase I is influenced by Mg²⁺ and Ca²⁺ but let the Binding of DNase I ions is still unclear, the active side is found, the amount and mechanism of Ca²⁺ and Mg²⁺. The results of the review in this article indicate that DNase I is needed externally from experimental animals and simulations improve the binding reaction of Ca²⁺ and Mg²⁺ which is supported by some literature that uses crystallographic studies where Ca²⁺ ions can stabilize DNase I and also in the presence of Mg²⁺ ions that can accelerate the hydrolytic failure of the cation. The results of the review explain the role of cations collectively electrostatic matching between DNase I and DNA.

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Introduction

Deoxyribonucleic acid (DNA) is a polymer molecule consisting of small subunit DNase I is a part of the pancreatic endonuclease that has an important role in the catalysis and hydrolysis of double-stranded DNA¹. Several previous studies used in the treatment of pulmonary complications have shown catalytic activity and hydrolysis of DNase I². Overall, DNase I has an important role in DNA hydrolysis, providing nutrients at certain times in DNA, can perform protection and intervention in the process of DNA degradation, cleaning DNA from extracellular media and also the reaction of polymerized actin³. DNase especially bpDNase I has been studied extensively where DNase I has high physiological pH activity due to the presence of Mg²⁺ and Ca²⁺.

DNase I activity is 100-fold lower in buffered States and conditions containing only one type of Cation⁵. This is due to the presence of factors that can cause enzymatic reactions. Hence the importance of the divalent carbon bonds Ca²⁺ and Mg²⁺ for the structural integrity of DNase I⁶, where the Ca²⁺ oxidation protecting bpDNase I from proteolytic/degradation obtains two disulfide reduction bonds. Equilibrium dialysis with the support of spectroscopic studies concluded that bpDNase I there are two strong bonds, and some weak cations catalytic binding site.

Some previous studies suggest that the disulfide and calcium bonds bind the Active Side (Side 1 and side 2) so that it becomes a functional part of DNase I. Based on the study of X-ray structure and biochemical studies of DNase I is Ca²⁺. In a previous study with molecular

dynamics simulation (md) in an explicit solvent conducted on DNase I found that the presence of four Ca²⁺ and Mg²⁺ cations has an important role in optimizing the electro-static match between the enzyme and negatively charged DNA^{14,15}. Mg²⁺ binds to the catalytic side of DNase I which has a role in hydrolysis with the help of cations, so that this event can provide an understanding of the direct relationship between cations from the biological function of DNase

Both active sides of DNase I bind Ca²⁺ as a single unit to bind active sites I and II²⁰. Divalent Mg²⁺ Cation is also in the catalytic pocket of DNase I which has a role in hydrolysis. In catalytic site DNase I obtained from 1n1dk structure of chemically modified bpDNase I Complex. The active side of Mg²⁺ involves two histidines His 134 and His 252 that coordinate scissile phosphates that often require Mg²⁺ in catalytic activity. Important role divalent cations participate in DNA hydrolysis or to stabilize DNA groups near DNA

Content

Deoxyribonuclease (DNase I)

DNase I is a class of enzymes that catalyze the hydrolysis of DNA with mechanism and by cleaving the phosphodiester bonds of DNA and Ca²⁺ and Mg²⁺ reaction.²⁴ Encoded complexes by multiple genes and express with controlled by tissues, DNase is capable to mechanism of hydrolyzing DNA in both of intracellular and other mechanism extracellular spaces. This enzyme belongs to endonuclease pancreatic and can play a key role to determining and concentration of extracellular DNA (ecDNA).

There are nine high resolution X-Ray apoptotic endonucleases-,containing bpDNase I and listed on text (IANTN, 2A40, 2A42, 2A41, 2D1K, 3DNI and IATN) five of which belong to the DNase I group site I, namely DNase I site I, DNase I-like 1 (DNase X), DNase I-like 2, and DNase I-like 3 (DNase γ). Furthermore, there are three other endonucleases from the DNase I 3DNI and IATN from site catalytic stie II group including the all site III contains with Ca^{2+} but not Mg^{2+} . DNase I activity is 100-fold lower in buffered States and conditions containing only one type of Cation. This is due to the presence of factors that can cause enzymatic reactions. Hence the importance of the divalent carbon bonds Ca^{2+} and mg^{2+} for the structural integrity of DNase I, where the CA^{2+} oxidation protecting bpDNase I from proteolytic/degradation obtains two disulfide reduction bonds. Equilibrium dialysis with the support of spectroscopic studies concluded that bpDNase I there are two strong bonds, and some weak cations catalytic binding site.

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between cations from the biological function of DNase I.

All of the observation supported and was analyzed with analysis of bound Crysthallographic of water molecules. And therefor with bound water molecules observed with site I and site II.

DNase I Catalytic site.

DNase I (Figure 1) is a DNA-specific site I, II,II and IV

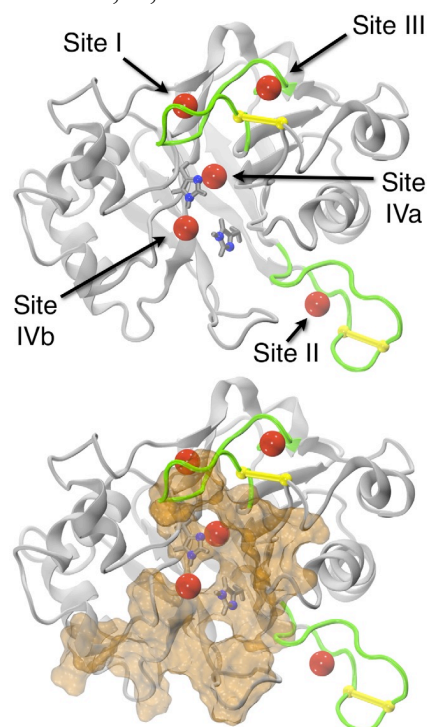


Figure 1. DNase I ion binding site

enzyme and is often used to study DNA and protein interactions with preservation crsthallography. DNase I is the most active endonuclease in DNA of mammalian cells, accounting for almost 100% of the total pancreatic endonuclease and activity in relevance and important organs, blood, and urine. Renal tubular epithelial cells and salivary glands express the highest amount of DNase I enzyme. Overall, these results demonstrate that bpDNase I has at least three possible

binding sites for divalent cations. The discovery of site III, in addition to the well-known sites I and II with strong affinity for Ca^{2+} , is particularly intriguing because it appears to preferentially bind Mg^{2+} . Verification of this site's ability to stabilize Mg^{2+} is still pending. In bpDNase I without actin. The presence of bound water molecules in three structures is the only evidence for the existence of a fourth putative site close to the bpDNase I catalytic pocket, which may be occupied by Mg^{2+} and must therefore be investigated from scratch. structural drift in simulations of bp DNase I The bp DNase I molecular dynamics simulations performed in the were used to further study the confirmed or potential cation-binding sites identified above.

Different kinds of metal ions are present in the solution. The initial placements of Na^{+} surrounding the enzyme in all seven simulations were established using a Coulombic potential grid and, crucially, were not situated in cation-binding sites. 2 provides a summary of the characteristics of the seven scenarios. According to the computed Ca-RMSD values between the free 3DNI bpDNase I and the simulated structures, which varied from 1.0 (Sim3) to 1.7 (Sim1) $60.1 \text{ \AA}^{\circ} (2)$, no significant structural reorganizations took place over the trajectories. The stability of these simulations is attested to by the small standard deviations (0.1 \AA°).

All generated structures, on the whole, stayed fairly close to the 3DNI crystal11,25 . structure. Number, area and coordination of cations in bpDNase I reenactments Locales I and II in bpDNase I, indistinguishable to their X-ray partners, were exceptionally solid cation-binding pockets, continuously possessed by

particles all through the directions . Na^{+} Suddenly possessed these locales within the nonappearance of Ca^{2+} (Sim1). Particle coordination was ideal for both Na^{+} and Ca^{2+} . Locales I and II were encourage fortified by authoritative Ca^{2+} (Figure 2), given its maximal coordination number (CNmax) of , higher than for Na^{+} (CNmax of). In expansion, compared to Na^{+} , the participation of Cys101 within the Cys101-Cys104 bridge fortified the amino corrosive side-chain coordination of Ca^{2+} at location II Both the amino-acid structure and the tricapped trigonal kaleidoscopic geometry of locales I and II are indistinguishable to those of crystallographic structures. Location III, close to location I (Figure 1), favored Mg^{2+} within the X-ray structures of bpDNase I in complex with actin

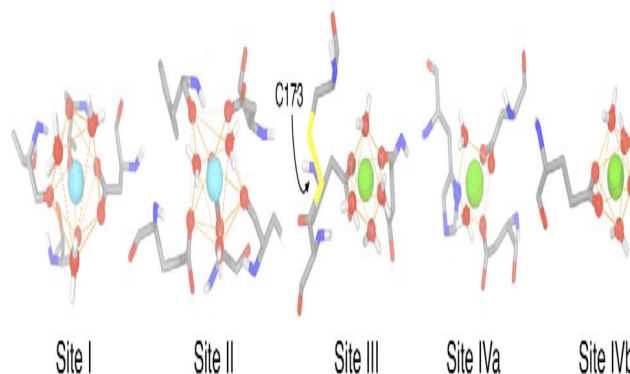


Figure 2. Coordination Spheres of Ca^{2+} and Mg^{2+} bound to DNase I

Basic impacts of particles at destinations I, II and III

Sites I and II are found in circles L1 (Leu195 to Tyr211) and L2 (Tyr97 to Pro113) individually (Figure 1). The auxiliary solidness of L1, reflected by the moo temperature variables in 3DNI and the low root cruel square vacillation (RMSF) values ($0.760.1 \text{ \AA}^{\circ}$ on average) in our

reenactments, was not influenced by the nature of the bound particle (Figure 3). Then again, the nearness of particles influenced L2, which was more adap than L1 (Figure 3). Na + in location II driven to higher L2-RMSF values (1.360.1 A ° on normal) than Ca2+ (0.960.1 A ° on normal). The reenactments encourage highlight that when Na + was bound at location II, the Gly100-Cys104 region significantly veered off from the setup watched with Ca2+ bound at this location (Figure 3). This proposes that Ca2+ at location II restricts L2 adaptability and stabilizes one of at slightest two possible conformations of L2. Mg2+ at location III was facilitated to Asp172, securing the Cys173-Cys209 bridge from diminishment. Appropriately, one water molecule included within the Mg2+ coordination circle was also interacting with Cys173 (Figure 2). This water atom, firmly bound by both Mg2+ and Cys173, did not trade with the solvent and in part protected the Cys173 sulfur iota, reducing its openness. This circumstance was particular to Mg2+ , since the coordination circle of Na + at location III was shaped by exchangeable water particles that seem not secure Cys173. Cation official at the already obscure location III gives a comprehensive interpretation of the resistance of the Cys173-Cys209 disulfide bridge to b-mercaptoethanol assault in a location I-defective bpDNase I variant [26]. Individually, cations stabilized the enzyme's structure, either directly (Ca2+ at location II) or in a roundabout way (Mg2+ at destinations III and IV).

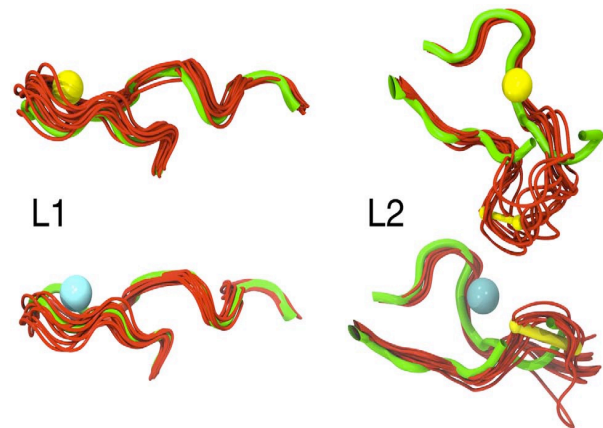


Figure 3. L1 dan L2 dynamics

References

- Weston SA, Lahm A, Suck D (1992) X-ray structure of the DNase Id(GGTATACC)2 complex at 2.3 A resolution. *J Mol Biol* 226: 1237–1256.
- Chen WJ, Lai PJ, Lai YS, Huang PT, Lin CC, et al. (2007) Probing the catalytic mechanism of bovine pancreatic deoxyribonuclease I by chemical rescue. *Biochem Biophys Res Commun* 352: 689–696.
- Dupureur CM (2008) Roles of metal ions in nucleases. *Curr Opin Chem Biol* 12: 250–255.
- Lahm A, Suck D (1991) DNase I-induced DNA conformation. 2 A structure of a DNase I-octamer complex. *J Mol Biol* 222: 645–667.
- Glazer DS, Radmer RJ, Altman RB (2009) Improving structure-based function prediction using molecular dynamics. *Structure* 17: 919–929.
- Lee SH, Kerff F, Chereau D, Ferron F, Klug A, et al. (2007) Structural basis for the actin-binding function of missing-in-metastasis. *Structure* 15: 145–155.
- Suck D, Oefner C (1986) Structure of DNase I at 2.0 A resolution suggests

- a mechanism for binding to and cutting DNA. *Nature* 321: 620–625.
- Bounds DG (1985) A molecular dynamics study of the structure of water around the ions Li⁺, Na⁺, K⁺, Ca⁺⁺, Ni⁺ and Cl⁻. *Molecular Physics* 54: 1335–1355.
- Jiao D, King C, Grossfield A, Darden TA, Ren P (2006) Simulation of Ca²⁺ and Mg²⁺ solvation using polarizable atomic multipole potential. *J Phys Chem B* 110: 18553–18559.
- Delalande O, Ferey N, Laurent B, Gueroult M, Hartmann B, et al. (2010) Multiresolution approach for interactively locating functionally linked ion binding sites by steering small molecules into electrostatic potential maps using a haptic device. *Pac Symp Biocomput.* pp 205–215.
- Babor M, Gerzon S, Raveh B, Sobolev V, Edelman M (2008) Prediction of transition metal-binding sites from apo protein structures. *Proteins* 70: 208–217.
- Pingoud A, Fuxreiter M, Pingoud V, Wende W (2005) Type II restriction endonucleases: structure and mechanism. *Cell Mol Life Sci* 62: 685–707.
- Sigel RK, Pyle AM (2007) Alternative roles for metal ions in enzyme catalysis and the implications for ribozyme chemistry. *Chem Rev* 107: 97–113.
- Yang W, Lee JY, Nowotny M (2006) Making and breaking nucleic acids: two-Mg²⁺-ion catalysis and substrate specificity. *Mol Cell* 22: 5–13.
- Pingoud V, Wende W, Friedhoff P, Reuter M, Alves J, et al. (2009) On the divalent metal ion dependence of DNA cleavage by restriction endonucleases of the EcoRI family. *J Mol Biol* 393: 140–160.
- Arcesi L, La Penna G, Perico A (2007) Generalized electrostatic model of the wrapping of DNA around oppositely charged proteins. *Biopolymers* 86: 127–135.
- Cherstvy AG (2009) Positively charged residues in DNA-binding domains of structural proteins follow sequence-specific positions of DNA phosphate groups. *J Phys Chem B* 113: 4242–4247.
- Nadassy K, Wodak SJ, Janin J (1999) Structural features of protein-nucleic acid recognition sites. *Biochemistry* 38: 1999–2017.
- Pan CQ, Lazarus RA (1998) Hyperactivity of human DNase I variants. Dependence on the number of positively charged residues and concentration, length, and environment of DNA. *J Biol Chem* 273: 11701–11708.
- Evans SJ, Shipstone EJ, Maughan WN, Connolly BA (1999) Site-directed mutagenesis of phosphate-contacting amino acids of bovine pancreatic deoxyribonuclease I. *Biochemistry* 38: 3902–3909.
- Kunitz M (1950) Crystalline desoxyribonuclease; digestion of thymus nucleic acid; the kinetics of the reaction. *J Gen Physiol* 33: 363–377.
- Shack J, Bynum BS (1964) Interdependence of Variables in the Activation of Deoxyribonuclease I. *J Biol Chem* 239: 3843–3848.
- Pan CQ, Lazarus RA (1997) Engineering hyperactive variants of human deoxyribonuclease I by altering its

- functional mechanism. *Biochemistry* 36: 6624–6632.
- Case DA, Cheatham TE, 3rd, Darden T, Gohlke H, Luo R, et al. (2005) The Amber biomolecular simulation programs. *J Comput Chem* 26: 1668–1688.
- Wang JC, Cieplak P, Kollman P (2000) How well does a Restrained Electrostatic Potential (RESP) Model perform in Calculation conformation Energies of Organic and Biological Molecules. *J Comput Chem*. pp 1049–1074.
- Berendsen HJC, Postma WF, Van Gunsteren A, DiNola A, Haak JR (1984) Molecular dynamics with coupling to an external bath. *J Chem Phys* 81: 3684–3690.
- Van Gunsteren WF, Berendsen HJC (1977) Algorithms for macromolecular dynamics and constraint dynamics. *Mol Phys* 34: 1311–1327.
- Darden T, York D, Pedersen L (1993) Particle mesh Ewald: an N.log(N) method for Ewald sums in large systems. *J Chem Phys* 98: 10089–10092.
- Vriend G (1990) WHAT IF: a molecular modeling and drug design program. *J Mol Graph* 8: 52–56,29.
- Grossfield A (2005) Dependence of ion hydration on the sign of the ion's charge. *J Chem Phys* 122: 024506.
- Baker NA, Sept D, Joseph S, Holst MJ, McCammon JA (2001) Electrostatics of nanosystems: application to microtubules and the ribosome. *Proc Natl Acad Sci U S A* 98: 10037–10041.
- Hubbard SJ, Thornton JM (1993) 'NACCESS', Computer Program. Department of Biochemistry and Molecular Biology, University College London.
- Prabhu NV, Panda M, Yang Q, Sharp KA (2008) Explicit ion, implicit water solvation for molecular dynamics of nucleic acids and highly charged molecules. *J Comput Chem* 29: 1113–1130.
- Altschul SF, Madden TL, Schaffer AA, Zhang J, Zhang Z, et al. (1997) Gapped BLAST and PSI-BLAST: a new generation of protein database search programs. *Nucleic Acids Res* 25: 3389–3402.
- Thompson JD, Higgins DG, Gibson TJ (1994) CLUSTAL W: improving the sensitivity of progressive multiple sequence alignment through sequence weighting, position-specific gap penalties and weight matrix choice. *Nucleic Acids Res* 22: 4673–4680.
- Heinig M, Frishman D (2004) STRIDE: a web server for secondary structure assignment from known atomic coordinates of proteins. *Nucleic Acids Res* 32: W500–502.
- Humphrey W, Dalke A, Schulten K (1996) VMD: visual molecular dynamics. *J Mol Graph* 14: 33–38, 27–38.
- Saladin A, Fiorucci S, Poulain P, Prevost C, Zacharias M (2009) PTools: an opensource molecular docking library. *BMC Struct Biol* 9: 27
- Team RDC (2006) A language and environment for statistical computing. Available at R Foundation for Statistical Computing, <http://www.R-project.org> [Internet]. 2011;19(1)