

Research Article

The effectiveness of analytical chemistry problem based learning model on student learning outcomes

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| Analytical chemistry Learning improvement results PBL model Learning outcomes Chemistry are subjects carried out in this study were 30 Tanjung The method used is to analyze the pre-test and post-test as we distribution of the G factor. From the results of the study it can The value of the pre-test results showed that out of 30 stu (3.33%) with the highest score with a score of 74. For 29 peop not to have passed because the score was below 74. 2). The 6.33 with a variance of 40.110. 3). the posttest score of 5 peop 90 with a percentage of 16.67%, and 83.33% showed a score PBL Chemistry Analysis lesson. The conclusion from the result outcomes in the analytical chemistry course was obtained that to | ents in learning analytical galai Polytechnic students. I as data on the frequency be concluded as follows 1). dents there was 1 person the (97.67%) were declared tandard deviation value is e with the highest score of below 90 and passed the stof the increased learning the g factor value was 0.56 |
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Introduction

The learning process requires innovative thinking with the aim of increasing the level of quality education and is supported by the government and several parties in the education sector. Teaching materials are one of the basics for developing quality education to improve the quality of understanding in learning the teaching materials taught by the lecturer (Simbolon, 2022). To achieve success it needs support from the government to improve the quality of the curriculum and teaching materials (Harahap et al. 2022).

The latest form of learning development is the Problem Based Learning Model which applies student activity in solving problems and can be done independently or in groups through predetermined stages and times (Education and Culture, 2018). The word learning is a system that is neatly arranged to carry out the teaching and learning process for students in which there is a learning design that affects internal teaching learning outcomes to produce learning outcomes (Hermawansa, 2021). And the form of learning outcomes is a form of achievement from the learning process that is carried out where students are able to carry out and show changes in terms of knowledge. attitudes and ways of thinking of students (Sriamah et al. 2020). Learning strategies have an important role in learning analytical chemistry. For this reason, lecturers improve the theory and skills learning system in the form of practice as problem solving. So, the basis for choosing learning must be observed as one of the analytical chemistry learning strategies that will be able to improve students' problem-solving abilities is the Problem-Based Learning Model (PBL) in Table 1 (Simamora et al. 2017; Purba et al. 2018; Nasution et al. 2019).

The advantages of PBL include (1) training students to use reasoning in problem solving; (2) training participants to make hypotheses in problem solving based on simple business concepts; (3) training critical and



contextual thinking skills with real business problems encountered; and (4) training students to conduct trials in proving the hypothesis (Dewi, 2022; Nisa and Silaban, 2022). The PBL model is used in this study as a solution to solving analytical chemistry problems by using an approach in learning that helps students find problems from learning analytical chemistry in theoretical and practical learning so that they can find strategies that they have determined to make a solution decision (Lukman et al. 2019; Muti'ah, 2021; Nisa et al. 2022). After that, the problem will be presented in practice (Antara, 2022).

 Table 1. Syntax of the Problem Based Learning model (modified fron Simamora et al. 2017).

| Phases | Teacher Activity | Students Activity | |
|--|--|--|--|
| <i>Phase 1</i> Student orientation to the problem | Explain the purpose of learning. Motivate students to be actively involved in solving the selected problem | Be able to understand the scope of chemistry and chemical analysis and their role in learning theory and practice. Can innovate in practice to find answers to a problem. | |
| <i>Phase 2</i> Organize students | Help students define and organize learning tasks related to the problem | Can limit and classify a problem by solving it with a specific method | |
| <i>Phase 3</i> Individual and group research guide | Encourage students to gather appropriate Information to carry out experiments for explanations and problem solving | Investigate the issues presented using the techniques employed to find solutions. | |
| <i>Phase 4</i> Develop and present the work | Assist students in planning and preparing suitable works such as reports, models and sharing assignments with friend | Prepare reports on analytical chemistry practice with the results of research conducted on each student using different methods | |
| <i>Phase 5</i> Analyze and evaluate the problem-solving proces | • Evaluate learning outcomes about the material that has been studied/ ask group presentation of the work | The results of assignments and practice reports can be accounted for through presentations to lecturers | |

Learning about chemistry is a science that is closely related to the composition. structure and properties. changes. dynamics. and energetics of substances that exist in human life at the high school level (Pratama et al. 2017). Basically it has been applied which involves skills and reasoning to understand the chemistry (Lusiyana et al. 2019). Chemistry learning can be said to be a form of science category (Arifin, 2021). Where in learning begins with the basic concepts of science. develops. applies and draws conclusions from science learning (Susi and Yenti, 2020). However, in reality, what often happens in the field is still using conventional methods and explaining the material according to the module. So that these conditions make students less trained in understanding and developing the learning and it is difficult to be able to analyze and apply it in the form of chemical concepts in everyday life (Sandabunga and Anwar, 2021).

Learning analytical chemistry courses for the student level which is divided into theory and practice. of course. requires a learning development process to be able to harmonize theory and practice (Yusfiani, 2020). In this study, it is considered necessary to do which aims to see the extent to which students' improvement in analytical chemistry learning can be applied in theory learning and practicum for 1 academic year.

Method

General procedure

Instruments of data collection techniques taken in this study used the form of Pre-Test (Pre-Test) and Post-Test (Post-Test) questions. And then conduct interviews to students and also to teachers. The data analysis technique used descriptive statistical analysis. Descriptive Statistical Analysis is used to describe the characteristics of the score of the research sample.

Data analysis

Techniques Data analysis techniques in research conducted using Excel software. The data analysis in this study includes: (1) the value of the pretest and postet frequency distribution data for students; and (2) the value of increasing student learning outcomes using the N-Gain score formula (G factor). For the g factor formula used is as follows:

$N - Gain = \frac{Posttest \ score \ - \ Pretest \ score}{Ideal \ score \ - \ Pretest \ score} \tag{1}$

For the distribution of factor scores and the value of the effectiveness category is listed in Table 2.

| Distribution of Gain Score | | | | | |
|-------------------------------|----------------------------|-----------------|--|--|--|
| Criteria I | g > 0.7 | High | | | |
| | 0.3 <u>< g <</u> 0.7 | Medium | | | |
| | g < 0.3 | High | | | |
| Category Effectiveness N-Gain | | | | | |
| Criteria II | < 40 | Not Effective | | | |
| | 40 - 55 | Less Effective | | | |
| | 56 - 75 | Quite Effective | | | |
| | > 76 | Effective | | | |
| | | | | | |

 Table 2. The score factor value and effectiveness category (Meltzer, 2002)

Results and Discussion

The results of the research obtained from pre-test data and post-test data of students in the analytical chemistry course at the Tanjungbalai Polytechnic Campus. Based on the results of calculations from these data can be described as follows.

Pre-test result data

For the pretest results obtained before the application of learning is carried out in the analyst chemistry course at the Tanjungbalai Polytechnic Fisheries Product Processing Engineering Study Program. The results in Table 3 below explain that the frequency distribution value of the pretest value in the Chemical Analysis course.

| Intervals | Frequency | Percent | Valid Percent | Cumulative Percent | Result | Value |
|-----------|-----------|---------|---------------|--------------------|----------|--------|
| 45 | 2 | 6.67 | 6.67 | 6.67 | N | 30 |
| 55 | 2 | 6.67 | 6.67 | 13.34 | Min | 45 |
| 56 | 2 | 6.67 | 6.67 | 20.00 | Max | 74 |
| 60 | 7 | 23.33 | 23.33 | 43.34 | Range | 29 |
| 62 | 1 | 3.33 | 3.33 | 46.67 | К | 5.907 |
| 63 | 1 | 3.33 | 3.33 | 50.00 | Р | 4.909 |
| 64 | 5 | 16.67 | 16.67 | 66.67 | Std. Dev | 6.33 |
| 65 | 6 | 20.00 | 20.00 | 86.67 | Variance | 40.110 |
| 66 | 1 | 3.33 | 3.33 | 90.00 | | |
| 68 | 1 | 3.33 | 3.33 | 93.34 | | |
| 73 | 1 | 3 33 | 3 33 | 96.67 | | |
| 74 | 1 | 3.33 | 3.33 | 100.00 | | |

Table 3. Frequency distribution value of pretest value of chemical analysis course

Table 3 presents the pretest values of experimental class students as comparisons before the application of the PBL learning model to do the essay test with 10 questions. From the results of the pre-test. the results showed that from 30 students there was 1 person (3.33%) with the highest score with a score of 74. For there were 29 people (97.67%) who did not pass because the score was below 74. The explanation in table 2 is contained in the form of Fig. 1.

Post test data

For the pretest scores obtained before the application of learning is carried out in the analyst chemistry course at the Fishery Product Processing Engineering Study Program. Tanjungbalai Polytechnic. The results in Table 4 below explain that the value of the frequency distribution of values posttest in chemical analysis course.

In Table 4 above explains that the posttest scores carried out from the experimental class present posttest scores for experimental class students after the application of the PBL learning model did an essay test with 10

questions and got 5 people with the highest score 90 with a percentage of 16.67% and 83.33% showed a value below 90. The standard deviation of the posttest value was 4.138. The explanation in table 2 is contained in the form of Fig. 2.

| Table 4. Frequency distribution of posttest scores for chemical analysis courses | | | | | | |
|--|-----------|---------|------------------|-----------------------|----------|--------|
| Intervals | Frequency | Percent | Valid Percent | Cumulative Percent | Result | Value |
| 76 | 1 | 3.33 | 3.33 | 3.33 | N | 30 |
| 78 | 1 | 3.33 | 3.33 | 6.66 | Min | 76 |
| 79 | 1 | 3.33 | 3.33 | 10.00 | Max | 90 |
| 80 | 7 | 23.33 | 23.33 | 33.33 | Range | 14 |
| 82 | 5 | 16.67 | 16.67 | 50.00 | ĸ | 5.907 |
| 83 | 1 | 3.33 | 3.33 | 53.33 | Р | 2.370 |
| 84 | 3 | 10.00 | 10.00 | 63.33 | St. Dev | 4.138 |
| 86 | 3 | 10.00 | 10.00 | 73.33 | Variance | 17.126 |
| 87 | 1 | 3.33 | 3.33 | 76.66 | | |
| 88 | 1 | 3.33 | 3.33 | 80.00 | | |
| 89 | 1 | 3.33 | 3.33 | 83.33 | | |
| 90 | 5 | 16.67 | 16.67 | 100.00 | | |



Figure 1. Graph of the pretest scores of experimental class students



Figure 2. Graph of posttest scores of experimental class students

Improved learning outcomes (N-gain score test)

To determine the acquisition of learning outcomes, the N-Gain factor formula is used in formula 1. For the results of the gain score and the effectiveness category, the value of the calculations carried out in accordance with Table 1 above can be described in Table 5. From the description in table 5 shows the value of the pretest and posttest processes, the g factor value is 0.56 in the medium category and the percentage is 56.16% in the quite effective category.

| N | Pre-Test | Post-test | Value of Post - Pre | Ideals Score | N-Gain Score | N-Gain Score Percen |
|-----|----------|-----------|------------------------|--------------|-----------------|---------------------------|
| 1 | 55 | 80 | 25 | 45 | 0.556 | 55 556 |
| 2 | 56 | 88 | 32 | 44 | 0.727 | 72 727 |
| 3 | 60 | 89 | 29 | 40 | 0.725 | 72.727 |
| 4 | 65 | 87 | 22 | 35 | 0.629 | 62 857 |
| 5 | 65 | 76 | 11 | 35 | 0.314 | 31 429 |
| 6 | 63 | 84 | 21 | 37 | 0.568 | 56.757 |
| 7 | 64 | 90 | 26 | 36 | 0.722 | 72.222 |
| 8 | 60 | 80 | 20 | 40 | 0.500 | 50.000 |
| 9 | 60 | 90 | 30 | 40 | 0.750 | 75.000 |
| 10 | 64 | 80 | 16 | 36 | 0.444 | 44.444 |
| 11 | 62 | 80 | 18 | 38 | 0.474 | 47.368 |
| 12 | 64 | 79 | 15 | 36 | 0.417 | 41.667 |
| 13 | 73 | 80 | 7 | 27 | 0.259 | 25.926 |
| 14 | 45 | 82 | 37 | 55 | 0.673 | 67.273 |
| 15 | 74 | 82 | 8 | 26 | 0.308 | 30.769 |
| 16 | 65 | 90 | 25 | 35 | 0.714 | 71.429 |
| 17 | 64 | 90 | 26 | 36 | 0.722 | 72.222 |
| 18 | 56 | 90 | 34 | 44 | 0.773 | 77.273 |
| 19 | 60 | 86 | 26 | 40 | 0.650 | 65.000 |
| 20 | 64 | 84 | 20 | 36 | 0.556 | 55.556 |
| 21 | 45 | 82 | 37 | 55 | 0.673 | 67.273 |
| 22 | 65 | 83 | 18 | 35 | 0.514 | 51.429 |
| 23 | 60 | 82 | 22 | 40 | 0.550 | 55.000 |
| 24 | 66 | 80 | 14 | 34 | 0.412 | 41.176 |
| 25 | 65 | 86 | 21 | 35 | 0.600 | 60.000 |
| 26 | 65 | 86 | 21 | 35 | 0.600 | 60.000 |
| 27 | 68 | 80 | 12 | 32 | 0.375 | 37.500 |
| 28 | 60 | 78 | 18 | 40 | 0.450 | 45.000 |
| 29 | 60 | 82 | 22 | 40 | 0.550 | 55.000 |
| 30 | 55 | 84 | 29 | 45 | 0.644 | 64.444 |
| Ανε | 61.600 | 83.667 | 22.067 | 38.400 | 0.562 Medium | 56.160 Quite Effective |

Conclusion

The pre-test results show that out of 30 students, there is 1 person (3.33%) with the highest score of 74. There are 29 people (97.67%) who do not pass because the score is below 74. And the standard deviation is 6.33 with a variant value. 40,110. And the value of the Post Test is the score of 5 people with the highest score of 90 with a percentage of 16.67% and 83.33% showing a score below 90 and passing the PBL analysis chemistry lesson. The standard deviation value for the posttest is 4.138 with a variance of 17.126. While the

value of increasing learning outcomes in analytical chemistry courses. the g factor value of 0.56 is in the moderate category and the percentage is 56.16% which is quite effective.

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